

Radiation-induced Phenylation of *p*-Bromophenol in Aqueous Solution

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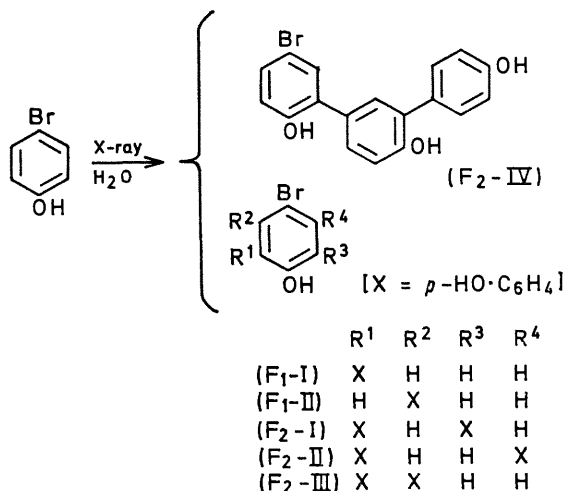
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Summary New hydroxy-biphenyls and -terphenyls were formed by radiolysis of *p*-bromophenol in aqueous solution.

WHILE studying the γ -radiolysis of *p*-bromophenol in aqueous solution, we have observed the formation of new

phenol oligomers. These oligomers may be formed by the arylation of *p*-bromophenol or of the dimeric product by aryl radical intermediates, resulting from the debromination of *p*-bromophenol by some radiolysis product(s) of water.¹ This communication deals with the identification of these oligomers, two asymmetric biphenyls and four terphenyls.

The *p*-bromophenol solution (25 mM) in triply distilled water was irradiated under nitrogen gas at room temperature with γ -rays from a ^{60}Co source; the total doses were 1.5 Mrad. From the ether extracts of the irradiated



p-bromophenol solution, several phenolic fractions (Folin-Denis reagent positive) were separated by gradient elution chromatography in benzene-ethyl acetate. Fraction 1 was rechromatographed to give 5-bromo-2,4'-dihydroxybiphenyl (F₁-I); m.p. 134°; λ_{max} (EtOH) 224, 254, and 299 nm (log ϵ 4.3, 4.0, and 3.8). The structure assigned to (F₁-I) is supported by the n.m.r. spectrum of the methyl ether which showed the presence of two methoxy-groups (δ 3.75, 3H, s and 3.69, 3H, s) and three protons *ortho* to the methoxy-group (δ 6.71, 1H, d *J* 8 Hz and 6.81, 2H, d *J* 8 Hz). The structure was further confirmed by the synthesis of 5-bromo-2,4'-dimethoxybiphenyl² using the method of Hodgson *et al.*³ The presence of 2-bromo-5,4'-dihydroxybiphenyl (F₁-II) in fraction 1 was evident from g.l.c. data on synthetic 2-bromo-5,4'-dihydroxybiphenyl silyl ether.²

The methylated fraction 2 gave four peaks on g.l.c., and these products could be isolated by column chromatography. All these compounds have the same molecular formula $\text{C}_{21}\text{H}_{19}\text{O}_3\text{Br}$; $M = 399$ (from mass spectra). Trimethyl ethers of (F₂-I), (-II), and (-III) were assigned as *meta*-, *para*-, and *ortho*-terphenyls, respectively, in which C-4, C-2', and C-4'' are substituted by methoxy-groups and C-5' by a bromine atom. The trimethyl ether of (F₂-IV) is 5-bromo-2,4,4'-trimethoxy-*m*-terphenyl. The evidence for the structures is as follows; (F₂-I) trimethyl ether (m.p. 94°): λ_{max} (EtOH) 263 and 302(sh)nm. (log ϵ 4.54 and 3.70); n.m.r. (CCl_4) δ 3.08, 3H, s and 3.84, 6H s (ArOCH_3), 6.88, 4H, d *J* 9 Hz (protons *ortho* to methoxy-groups), 7.49, 4H, d *J* 9 Hz (*ortho* coupling ArH), 7.35, 2H, s (isolated ArH); (F₂-II) trimethyl ether (m.p. 169—170°): λ_{max} (EtOH) 224(sh), 275, and 306 nm (log ϵ 4.32, 4.27, and 3.98); n.m.r. (CCl_4) δ 3.72, 3H, s, 3.77, 3H, s and 3.78, 3H, s (ArOCH_3), 6.78, 1H, s (isolated proton *ortho* to methoxy-group), 6.82, 2H, d *J* 9 Hz and 6.85, 2H, d *J* 9 Hz (protons *ortho* to methoxy-groups), 7.41, 1H, s (isolated ArH), 7.35, 2H, d *J* 9 Hz, and 7.28, 2H, d *J* 9 Hz (*ortho* coupling ArH); (F₂-III) trimethyl ether (m.p. 121—122°): λ_{max} (EtOH) 229, 239 (sh), 280, and 285 (sh) nm (log ϵ 4.44, 4.41, 3.93, and 3.90); n.m.r. (CCl_4) δ 3.68, 3H, s, 3.70, 3H, s, and 3.71, 3H, s (ArOCH_3), 7.53, 1H, d *J* 9 Hz (*ortho* coupling proton deshielded by bromine atom) and nine *ortho*-protons including four aromatic protons shielded by nonplanarity of benzene rings and five protons *ortho* to methoxy-group (δ 6.59, 2H, d *J* 9 Hz, 6.65, 2H, d *J* 9 Hz, 6.78, 1H, d *J* 9 Hz, 6.82, 2H, d *J* 9 Hz, and 6.87, 2H, d *J* 9 Hz); (F₂-IV) trimethyl ether (m.p. 130—131°): λ_{max} (EtOH) 223, 256, and 295 nm (log ϵ 4.47, 4.48, and 4.11); n.m.r. (CCl_4) δ 3.79, 3H, s and 3.83 6H, s (ArOCH_3), 6.80, 1H, d *J* 9 Hz, 6.89, 2H, d *J* 9 Hz, and 6.93, 1H, d *J* 9 Hz (aromatic protons *ortho* coupling to methoxy-groups); six other aromatic protons appeared at lower field than 7.0 p.p.m. The *ortho*-, *meta*-, and *para*-structures of the four terphenyls are in agreement with the u.v. spectroscopic data for *o*-, *m*-, and *p*-terphenyl, λ_{max} (EtOH) 237.5, 250.9 and 273.7 nm, respectively.⁴

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